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Mechanism of Heterogeneous Catalysis Clarified by Surface Science Technique

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Heterogeneous catalysts are pivotal materials in chemical industry. However, the mechanism of catalytic reactions taking place on solid surfaces is very complex. Direct observation of catalyst surface by surface-science technique is needed to clarify the mechanism of catalytic reactions and the active catalytic sites. The most important point in the surface science approach is to establish a model catalyst with well-defined structures, in which the catalytic sites must be identical to those of industrial catalysts. In my presentation, model catalyst studies will be introduced concerning methanol synthesis from CO₂ and nitrogen-doped carbon catalysts for fuel cells. In the first topic, I will be talking about CuZn active sites of Cu/ZnO catalysts for the methanol synthesis from CO₂, which has been a controversial issue¹⁾. A unique Eley-Rideal typed mechanism of CO₂ with adsorbed hydrogen on Cu surfaces will be also introduced²⁾, which has been accidentally discovered during the methanol synthesis study. In the second topics, I will introduce the model study of nitrogen doped carbon catalysts that clarifies which nitrogen, graphitic nitrogen or pyridinic nitrogen creates the active site of oxygen reduction reaction (ORR)³⁾.



Fig.1 Reaction takes place by supplying vibrational energy only to CO₂ without heating a Cu catalyst. *Nature Chemistry* (ref.2)

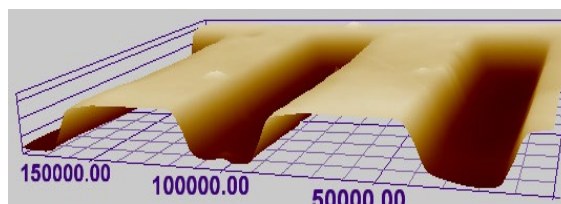


Fig.2 Active sites of nitrogen-doped carbon catalysts are created by pyridinic nitrogen which has been clarified by graphite model catalysts. *Science* (ref.3)

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- 2) J. Quan, F. Muttaqien, T. Kondo, T. Kozarashi, T. Mogi, T. Imabayashi, Y. Hamamoto, K. Inagaki, I. Hamada, Y. Morikawa, J. Nakamura, *Nature Chem.* **2019**, 11,722.
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PROFILE

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